FUNCTIONALISATION OF 25-METHYL GROUP OF D.A-FRIEDOOLEANAMES

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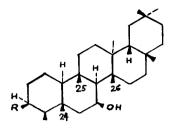
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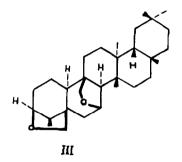
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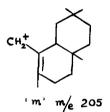
Although a number of triterpenoids of D:A-friedcoleanane group with oxygen function at C-25 have been isolated and their structures suggested¹⁻⁵, direct correlation with a triterpene of known structure and configuration has not been possible yet for the lack of a suitable method for introducing an oxygen function at C-25 methyl group. The present communication describes the first successful method for the functionalisation of the C-25 methyl group from known precursors friedcolean-76-ol (Ia)⁶⁻⁸ and 36-acetoxy friedcolean-76-ol (Ib)⁶⁻⁸.

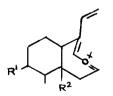
The axial 7β -hydroxyl group in both Ia and Ib suffers from severe 1,3-diaxial interaction by three methyl groups at C-24, C-25 and C-26, and hence it was suspected that hypohalite oxidation⁹ might effect functionalisation of any or all of the three methyl groups. Irradiation of a mixture of Ib, lead tetraacetate, iodine and calcium carbonate in cyclohexane for 3.5 hrs with a 500 watt tungsten lamp furnished a gummy mixture which could be resolved into three components on chromatography over alumina. The least polar fraction (5%), m.p. 274-276° was identified as <u>epi-friedelanol acetate</u> (I.R., TLC and mixed mp), presumably formed by



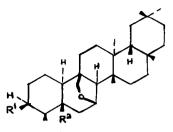
1 a, R≃H b, R=OAC



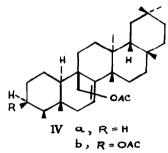


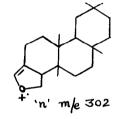


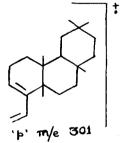
 $^{\circ}O' R^{1}=H, R^{\frac{1}{2}}CH_{8}, m/e 205$ $R'=OAC, R^{\frac{1}{2}}CH_{8}, m/e 263$ $R'=OAC, R^{2}=CH_{2}I_{3}m/e 389$



II a, $R^{I} = H$, $R^{2} = CH_{3}$ b, $R^1 = OAC$, $R^2 = CH_3$ C, $R^1 = OAC$, $R^2 = CH_2I$







hydrogen radical exchange with cyclohexane.

The second component $C_{32}H_{52}O_3$ to come out of the column, m.p. 215-218°, (α)_D + 44.7° (50%), M⁺ 484, <u>IR</u> 1740, 1250 (acetate) and 842 and 850 cm⁻¹ (ether); <u>MMR</u>: 0.85 (3H, d, J=4 Hz), 0.99 (3H, s), 1.05 (3H, s), 1.09 (3H, s), 1.12 (3H, s), 1.30 (3H, s) and 1.39 (3H, s) for one secondary and six tertiary methyl groups; 2.02 (3H, s, -CO.CH₃); AB quartet at 3.49 and 3.78 (2H, J=4 Hz, -CH₂-0- at C-25); 4.29 (1H, m, H at C-7) and 4.92 (1H, m, H at C-3) was attributed structure (IIb) on the basis of these data coupled with mass spectrometric fragmentation pattern described below.

The most polar component, $C_{32}H_{51}O_{3}I$, m.p. 123-125° contained iodine and was attributed structure (IIc), <u>IR</u> 1725, 1230 (acetate) and 842 (ether); <u>NMR</u> six methyl signals in the region 0.88 to 1.2, 2.18 (3H, s, -CO.CH₃), 2.64 and 2.84 (2H, AB quartet, J=4 Hz, shielded O-CH₂-), 3.62 (2H, s, -CH₂.I), 4.25 (2H, broad m, overlapped C-7 and C-3 H). On treatment with LiAlH₄ IIc afforded the diether (III), $C_{30}H_{48}O_2$, m.p. 221-225°, M⁺ 440, <u>IR</u> no acetate peak, peak at 850 cm⁻¹ (ether), <u>NMR</u> six methyl signals in the region 0.82 to 1.28; 3.42 and 3.61 (2H, AB quartet, J=2 Hz, -O-CH₂- at C-25), 3.84 (2H, broad, -O-C₁₂- at C-24), 4.09 -4.52 (2H, m, overlapped C-7 and C-3 H).

Similar irradiation of Ia in presence of lead tetracetate and iodine furnished after chromatography only two products. The less polar component (10%) was friedelane, m.p. 245-246°. The more polar component (60%), $C_{30}H_{50}O$, m.p. 204-206°, K⁺ 426, <u>IR</u> 840 and 850 cm⁻¹ (ether) was attributed structure (IIa) on the basis of mass spectral data.

The mass spectra¹⁰ of each of IIa, IIb and IIc showed a peak at m/e 205 corresponding to the ion 'm', thus demonstrating that the 0-26 methyl group had not been affected, A peak at m/e 302 from each of IIa,

IIb and IIc corresponded to ion 'n'^{ll} and definitely established that functionalisation had taken place at C-25. This conclusion was further supported by the appearance of a peak at m/e 205 from IIa, at m/e 263 from IIb and at m/e 389 from IIc corresponding to the ion '0'.

IIb on mild treatment with BF_3 -etherate in acetic anhydride gave the en-acetate (IVb), $C_{34}H_{54}O_4$, m.p. 160-162°, M⁺ 526, MMR 2.0 (6H, s, -CO.CH₃ at C-3 and C-25), 4.84 (1H, broad, -C=C-H at C-7). Similar treatment of IIa furnished IVa, m.p. 122-126°, M⁺ 468. The mass spectra of each of IVa and IVb gave a prominent peak at m/e 301 corresponding to the ion 'p' formed by RDA cleavage followed by elimination of -CH₂.0Ac.

All the compounds gave fairly good C,H-analytical data.

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